

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
15 March 2001 (15.03.2001)

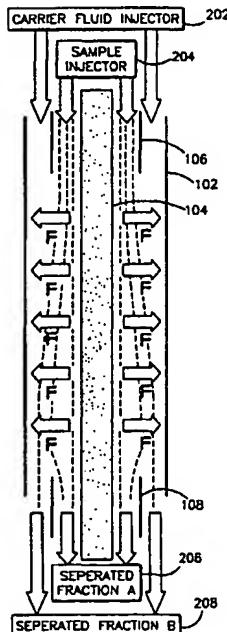
PCT

(10) International Publication Number
WO 01/17662 A2

- (51) International Patent Classification⁷: **B01D 49/00**,
69/02, 63/06 STATE UNIVERSITY [US/US]; 1960 Kenny Road,
Columbus, OH 43210 (US).
- (21) International Application Number: PCT/US00/24310 (72) Inventors; and
(75) Inventors/Applicants (*for US only*): ZBOROWSKI, Ma-
(22) International Filing Date: ciej [PL/US]; 541 Parkside Drive, Bay Village, OH 44140
1 September 2000 (01.09.2000) (US). CHALMERS, Jeff [US/US]; 4539 Tetford Road,
(25) Filing Language: English Columbus, OH 43220 (US). MOORE, Lee, R. [US/US];
(26) Publication Language: English 3344 Hyde Park Avenue, Cleveland Hts, OH 44118 (US).
(30) Priority Data: (74) Agent: PEJIC, Nenad; Calfee, Halter & Griswold LLP,
60/152,303 3 September 1999 (03.09.1999) US 800 Superior Avenue, Suite 1400, Cleveland, OH 44114
(US).
(71) Applicants (*for all designated States except US*): THE (81) Designated States (*national*): AE, AG, AL, AM, AT, AU,
CLEVELAND CLINIC FOUNDATION [US/US]; 9500 AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ,
Euclid Avenue, Cleveland, OH 44195 (US). THE OHIO DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,
HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ,

[Continued on next page]

(54) Title: CONTINUOUS PARTICLE AND MOLECULE SEPARATION WITH AN ANNULAR FLOW CHANNEL



(57) Abstract: The present invention provides apparatuses and methods for continuously separating particles of interest from a heterogeneous particle population. The heterogeneity can be based on, for example, magnetic susceptibility, particle size, thermal diffusion, phase solubility, and combinations of the preceding. Based on the heterogeneity, a separation force capable of exploiting the separand (i.e., particles subject to the separation process) is provided. The various embodiments of the present invention preferably employ an annular separation channel, appropriate separation force and flow compartments. In a first embodiment, an annular separation channel having semi-permeable inner and outer cylindrical walls is used to generate lateral convection forces. In

WO 01/17662 A2

[Continued on next page]



NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM,
TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

Published:

— *Without international search report and to be republished upon receipt of that report.*

- (84) **Designated States (regional):** ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

a second embodiment of the present invention, an annular separation channel having heat conductive inner and outer cylindrical walls is used to generate thermal diffusion forces. In a third embodiment, an annular separation channel having non-permeable inner and outer cylindrical walls is used to generate a solubility difference separation force based on solubility coefficient differences between phases of flow compartments. In a fourth embodiment, an annular separation channel having electrically conductive inner and outer cylindrical walls is used to generate an electrophoretic separation force. In a fifth embodiment, an annular separation channel having electrically conductive inner and outer cylindrical walls is used to generate a dielectrophoretic separation force. Yet other embodiments of the present invention provide for the combination of separation forces including lateral convection, thermal diffusion, solubility difference, electrophoretic, dielectrophoretic, and magnetic separation forces.

CONTINUOUS PARTICLE AND MOLECULE SEPARATION WITH AN ANNULAR FLOW CHANNEL

Field of the Invention

The invention relates generally to a device and method for continuously sorting particles and, more particularly, to a device and method for sorting particles through the use of an annular flow chamber and one or more types of separation forces.

Background of the Invention

Generally, particle separation devices separate particle populations of interest from a suspension and/or other types of particles. The principal method of operation of early particle separation devices relied on a particle's physical parameters to distinguish that particle from a suspension and/or other types of particles. Examples of these bulk separation techniques include filtration, which is based on particle size, and centrifugation, which is based on particle density. These techniques are effective as long as the particle population of interest is significantly different, with respect to size or density, from the suspension and/or the other particles in the population.

As a subset of bulk separating, continuous separation techniques also exist. The continuous separation of particles in flowing solution requires a well-defined and well-controlled fluid flow pattern. Typically, continuous particle separation devices employ rectangular separation channels. The rectangular geometry of such separation channels results in several advantages including, for example, ease of manufacture, ease of control of fluid flows inside the channels, and ease of design and implementation of forces that drive the separation.

However, rectangular separation channels also suffer from a drawback known as the sidewall effect. The sidewall effect distorts the fluid flow pattern at the side walls of the rectangular separation channel and, hence, adversely affects the performance of the sorting device such as, for example, its resolving power. Hence, it is highly desirable to provide methods and devices for separating particles that do not suffer from sidewall effects and can employ any one of a diverse number of separation forces.

Summary of the Invention

The present invention is particularly directed to continuous particle and molecule separation in flowing solutions. More particularly, the present invention addresses the issue of the sidewall effect by wrapping a separation channel around a cylinder running 5 the length of the channel and by joining the sides of the channel to thereby eliminate any side walls. So configured, the separation channel has an annular cross-section that is enclosed by two coaxial cylinders. Such a transformation leaves the length of the separation channel unchanged with the width of the channel laid on the cylinder. The separation channel width is equal to the difference between the radii of the two cylinder 10 walls bounding the separation channel.

In addition, the present invention generates axially-symmetric separation forces that are coaxial with the annular separation channel to continuously separate particles of interest from a heterogeneous particle population. The heterogeneity can be based on, for example, magnetic susceptibility, particle size, thermal diffusion, phase solubility, and 15 combinations of the preceding. Based on the heterogeneity, a separation force capable of interacting with the separand (*i.e.*, particles subject to the separation process) is provided. The particles or separands can be non-organic (*e.g.*, metals) or organic (*e.g.*, cells, viruses, and molecules including proteins and DNA). The various embodiments of the present invention preferably employ an annular separation channel, an appropriate 20 separation force, and flow compartments. In a first embodiment, an annular separation channel having semi-permeable inner and outer cylindrical walls is used to generate lateral convection forces. In a second embodiment of the present invention, an annular separation channel having heat conductive inner and outer cylindrical walls is used to generate thermal diffusion forces. In a third embodiment, an annular separation channel 25 having non-permeable inner and outer cylindrical walls is used to generate a solubility difference separation force based on solubility coefficient differences between phases of flow compartments. In a fourth embodiment, an annular separation channel having electrically conductive inner and outer cylindrical walls is used to generate an electrophoretic separation force. In a fifth embodiment, an annular separation channel 30 having electrically conductive inner and outer cylindrical walls is used to generate a

dielectrophoretic separation force. Yet other embodiments of the present invention provide for the combination of separation forces including lateral convection, thermal diffusion, solubility difference, electrophoretic, dielectrophoretic, and magnetic separation forces.

5

Brief Description of the Drawings

In the accompanying drawings which are incorporated in and constitute a part of the specification, embodiments of the invention are illustrated, which, together with a general description of the invention given above, and the detailed description given below, serve to example the principles of this invention.

10 Figure 1 is a schematic view of the annular separation channel of the present invention.

Figure 2 is a longitudinal section of the annular separation channel of the present invention along section lines 2-2 of Figure 1 and additionally showing carrier and sample injectors and separated fraction outputs

15 Figure 3 is a cross-sectional view of an embodiment of the annular separation channel of Figure 1 illustrating lateral convective separation.

Figure 4 is a cross-sectional view of an embodiment of the annular separation channel of Figure 1 illustrating thermal diffusion separation.

20 Figure 5 is a cross-sectional view of an embodiment of the annular separation channel of Figure 1 illustrating solubility difference separation.

Figure 6 is a cross-sectional view of an embodiment of the annular separation channel of Figure 1 suitable for electrophoretic separation and dielectrophoretic separation.

Detailed Description of Illustrated Embodiment

Referring now to Figure 1, a schematic view of the annular separation channel 100 is shown. The channel 100 has an outer cylindrical wall 102 (sometimes referred to as an "accumulation wall"), an inner cylindrical wall 104, and first and second cylindrical flow splitters 106 and 108, respectively. While two flow splitters are shown, there may 5 be more flow splitters per channel than two.

So configured, the distribution of the flow rates between the carrier fluid injector 202, sample injector 204, separated fraction a 206, and separated fraction b 208, generate 10 flow compartments that are used to divide the separand flow into individual flow compartments. The physical position of the flow compartments can then be adjusted by varying any one or more of the flow splitter location (*i.e.*, radius) in the annular separation channel, and carrier medium and sample flow rates. The distribution of the separation forces (*i.e.*, magnitude and direction) across the cross-section of the annular channel 100, together with the position of the flow compartments, determine the results 15 of the separation. Flow compartments are more fully described in U.S. Patent No. 5,968,820 to Zborowski et al. titled "Method for Magnetically Separating Cells into Fractionated Flow Streams" (hereinafter Zborowski), which is hereby fully incorporated by reference.

Illustrated in Figure 2 is a longitudinal section of the annular separation channel 20 100, along with a schematic representation of carrier fluid injectors 202, sample fluid injector 204, and separated fractions "a" 206 and "b" 208. Figure 2 also illustrates a preferred system arrangement of the present invention for continuously separating particles with the various embodiments of the annular separation channel disclosed herein. Except as specifically noted in the embodiments to follow, the carrier fluid or 25 medium is preferably phosphate-buffered saline at a physiological concentration of 0.15 mole per liter or other similar medium. Sample injector 204 introduces separands or particles preferably suspended in the same or similar fluid as the carrier medium. As will be presently described, a feature of the present invention is the development and use of

radially directed forces "F," which may take the form of any one of a number of identities.

Referring now to Figures 2 and 3, an embodiment of the present invention employing a lateral convective force "F" is shown. The convective force "F" is generated by making the inner and outer cylindrical walls 102 and 104, respectively, from a semi-permeable membrane material which allows for the transfer of a carrier medium (*i.e.*, solvent) but not the particles to be separated (*i.e.*, separands). The semi-permeable membrane is preferably made of regenerated or modified cellulose, polyacrylonitrile, polysulfone or polycarbonate. The lateral transport of the carrier medium is driven by a pressure difference of $P_2 - P_1$ between the inside of the semi-permeable inner wall 104, which is filled with the carrier medium, and the outside of the semi-permeable outer wall 102, which is also filled with the carrier medium. Hence, the carrier medium or fluid is injected inside of the semi-permeable inner wall 104 to form a first pressure P_1 and outside of the semi-permeable outer wall 102 to form a second pressure P_2 , wherein first pressure P_1 is greater than second pressure P_2 . The sample of separands or particle is introduced near the inner wall 104, as shown in Figure 2. The radial flow of the carrier medium pushes the particles against the semi-permeable outer wall 102 and the resulting radial concentration causes a diffusive flux back towards the center (*i.e.*, towards semi-permeable inner wall 104.) At equilibrium, larger particles stay close to the outer cylindrical wall 102 and are eluted in fraction "b" 208, whereas smaller particles stay farther away from the accumulation wall (*i.e.*, outer wall 102) and are eluted as the fraction "a" 206. It should also be noted that a reverse radial convective transport force can be generated by providing a first pressure P_1 that is less than the second pressure P_2 .

Referring now to Figures 2 and 4, an embodiment of the present invention employing a thermal diffusion force "F" is shown. In this embodiment, the outer cylindrical wall 102 is made of a heat-conductive material in contact with a large capacity heat reservoir of temperature T_2 . The inner cylindrical wall 104 is made of a heat-conductive material in contact with a large-capacity heat reservoir of temperature T_1 . Suitable heat conductive materials include, for example, a copper rod with imbedded

heating elements for the heat source T1 (or T2), and a copper cylinder with imbedded cooling elements for the heat sink T2 (or T1, correspondingly). Sample injector 204 introduces separands or particles near the inner cylindrical wall 104 and carrier medium injector 202 introduces a carrier medium into the flow channel. A large temperature difference, $T_2 - T_1$, drives a thermal diffusion of particles along the radial direction shown. Particles that are characterized by a large thermal diffusion coefficient (such as, for example, certain polymers) are particularly affected. Polymers differing in their thermal diffusion coefficients are collected in different fractions "a" 206 and "b" 208 at the exit from the annular channel. If temperature T_1 is less than temperature T_2 , particles with a large thermal diffusion coefficient tend to be collected in fraction "b" 208 and particles with lesser or no thermal diffusion coefficient tend to be collected in fraction "a" 206.

Referring now to Figures 2 and 5, an embodiment of the present invention employing a phase separation force "F" is shown. In this embodiment, a thin annular channel facilitates continuous liquid-to-liquid extraction processes in which separands separate between two different liquid phases based on differences in the separands' solubility coefficients. One embodiment of a phase separation force is generated by the solubility difference between an aqueous phase and an organic phase of a mixture of polar and non-polar separands. The polar separands stay in the aqueous phase and the non-polar separands move to the organic phase. Such a partition is generated by introducing an organic solvent as a carrier medium 202 and a mixture of separands " C_i "(non-polar) and " C_j " (polar) in an aqueous solvent as the sample 204. Suitable organic solvents include ether and hexane. Providing that there is no mixing between the two flows (i.e., the flow is substantially laminar), and the flow rates are chosen so that the residence time inside the separation channel 100 is significantly longer than the diffusion time necessary for the separands to traverse the channel width, the organic phase becomes enriched with the non-polar separand " C_i ," with the final concentration of non-polar separands in the organic phase being " C_i^b ." The aqueous phase becomes enriched with the polar separand " C_j ," with the final concentration of polar separands in the aqueous phase being " C_j^a ." Assuming that the concentration of the non-polar separand

remaining in the aqueous phase is "C_i^a" and that the concentration of the polar separand in the organic phase is "C_j^b," a separation factor α can be characterized as follows:

$$\alpha = \left(\frac{C_i^b / C_i^a}{C_j^b / C_j^a} \right)$$

- The embodiments of Figures 1 through 5 may also be combined individually,
5 collectively, or otherwise combinatorially with magnetic particle separation methods. More specifically, the lateral convective separation illustrated in Figures 2 and 3 can be combined with the magnetic separation methods and apparatus described in Zborowski, which is hereby fully incorporated by reference. Zborowski describes, among other things, the use of an annular separation channel in combination with magnetic forces and
10 flow compartments to achieve particle separation. The thermal diffusion separation illustrated in Figures 2 and 4 and the solubility difference separation illustrated in Figures 2 and 5 can also be combined with the magnetic separation described in Zborowski.

- Referring now to Figures 2 and 6, an embodiment of the present invention employing either an electrophoretic or dielectrophoretic separation force "F" is shown.
15 In this embodiment, the outer cylindrical wall 102 is made of an electrically conductive material at an electric potential of V2 volts. The inner cylindrical wall 104 is made of an electrically conductive material at an electric potential of V1 volts. Suitable electrically conductive materials include copper, steel, or metal alloy typically used for electrode construction. The electrically conductive material may be separated by an insulator to
20 form a pattern on cylindrical walls 102 and 104 as shown in Figure 7. Examples of geometrical patterns used for electrode construction include interdigitated electrodes for dielectrophoresis. Sample injector 204 introduces separands or particles near the inner cylindrical wall 104 and carrier medium injector 202 introduces a carrier medium into the flow channel. An electric potential difference, V2 - V1, drives an electrically susceptible
25 particles along the radial direction shown. This effect is known as electrophoretic mobility. Particles that are characterized by a large electrophoretic mobility are particularly affected. Particles differing in their electrophoretic mobility are collected in

different fractions "a" 206 and "b" 208 at the exit from the annular channel. If electrical potential V1 is less than electrical potential V2, particles with large electrophoretic mobility tend to be collected in fraction "b" 208 and particles with lesser or no electrophoretic mobility factors tend to be collected in fraction "a" 206.

5 Figures 2 and 6 are also applicable to dielectrophoretic separation forces. The difference between electrophoretic separation and dielectrophoretic separation being that a varying electric field is used in dielectrophoretic separation, whereas a static electric field is used for electrophoretic separation. Therefore, the above discussion of electrophoretic separation is equally applicable to dielectrophoretic separation with the
10 exception of the varying electric fields as represented by varying potentials V1 and V2.

Referring now to Figure 7, a separation channel 700 particularly suitable for dielectrophoresis is shown. The channel 700 includes a plurality of electrodes on outer cylindrical wall 102 such as, for example, electrodes 714 and 716, and inner cylindrical wall 104 such as, for example, electrodes 710 and 712. An insulator 715 resides between
15 electrodes 714 and 716. Similarly, an insulator 711 resides between electrodes 710 and 712. The electrodes and insulators are preferably interdigitated as shown in Figure 7 with equal distances between the electrodes. However, the distances may also be non-equally spaced to either concentrate or dilute the generated electrical forces.

A first source providing varying electrical potential V1 is in circuit
20 communication with the inner cylindrical wall 104 electrodes. More specifically, positive terminal 702 and negative terminal 704 provide each pair of electrodes such as, for example, electrodes 712 and 710, respectively, with the varying electrical potential V1. This configuration of electrode connection is continued through the entire inner cylindrical wall 104 electrodes with the electrode next to electrode 712 being connected
25 to the opposite terminal. In this manner, each pair of electrodes are of opposite polarity and have an insulator therebetween. A second source providing varying electrical potential V2 is in circuit communication with the outer cylindrical wall 102 electrodes. In particular, positive terminal 706 and negative terminal 708 provide each pair of electrodes such as, for example, electrodes 716 and 714, respectively, with the varying

electrical potential V2. This configuration of electrode connection is also continued through the entire outer cylindrical wall 102 electrodes with the electrode next to electrode 716 being connected to the opposite terminal. In this manner, each pair of electrodes are of opposite polarity and have an insulator therebetween.

5 Additional advantages and modifications will readily appear to those skilled in the art. For example, the sorting methods and apparatuses of the present invention may also be applied to particles other than cells, such as cell organelles, viruses, inorganic particles, such as soil particles, and molecules such as proteins and DNA. The distribution of particles between sorted fractions can be controlled by adjusting the
10 properties of the carrier medium and/or the separation forces. For example, increasing or decreasing the flow rate of the carrier medium decreases or increases the amount of time a separand is within the annular separation channel. Changing the magnitude of the separation force or the distribution of magnitude of the separation force between the inner and outer cylindrical walls of the annular separation channel changes the distribution of
15 particles between sorted fractions. Therefore, the invention, in its broader aspects, is not limited to the specific details, the representative apparatus, and illustrative examples shown and described. Accordingly, departures may be made from such details without departing from the spirit or scope of the applicant's general inventive concept.

We claim:

- 1 1. A channel for separating particles comprising:
 - 2 (a) a first space at least partially bounded by a first wall, the first space
3 configured to bear a first pressure;
 - 4 (b) a second space at least partially bounded by a second wall, the second
5 space configured to bear a second pressure;
 - 6 (c) a third space at least partially bound by the first and second walls and
7 configured to allow the development of a convective force; and
 - 8 (d) the first and second walls comprising a semi-permeable fluid material.
- 1 2. The channel of claim 1 wherein the first wall is substantially cylindrical.
- 1 3. The channel of claim 1 wherein the second wall is substantially cylindrical.
- 1 4. The channel of claim 2 wherein the first wall further comprises a first radius and
2 the second wall further comprises a second radius.
- 1 5. The channel of claim 4 where the first radius is less than the second radius.
- 1 6. The channel of claim 1 further comprising at least one splitter located at least
2 partially within the third space.
- 1 7. The channel of claim 1 wherein the semi-permeable material comprises cellulose.
- 1 8. The channel of claim 1 wherein the semi-permeable material comprises
2 polyacrylonitrile.
- 1 9. The channel of claim 1 wherein the semi-permeable material comprises
2 polysulfone.

3 10. The channel of claim 1 wherein the semi-permeable material comprises
4 polycarbonate.

1 11. A channel for separating particles comprising:
2 (a) a first pressure means for bearing a first pressure;
3 (b) a second pressure means for bearing a second pressure; and
4 (c) a third pressure means for bearing a range of pressures between the first
5 pressure and second pressures that allows for the development of a convective force.

1 12. The channel of claim 11 wherein the third pressure means comprises a semi-
2 permeable fluid pressure means.

1 13. The channel of claim 11 wherein the first pressure is greater than the second
2 pressure.

1 14. A channel for separating particles comprising:
2 (a) a first space at least partially bounded by a first wall, the first space
3 comprising a first heat reservoir;
4 (b) a second space at least partially bounded by a second wall, the second
5 space comprising a second heat reservoir;
6 (c) a third space at least partially bound by the first and second walls and
7 configured to allow the development of a thermal diffusion force; and
8 (d) the first and second walls comprising a heat conductive material.

1 15. The channel of claim 14 wherein the first wall is substantially cylindrical.

1 16. The channel of claim 14 wherein the second wall is substantially cylindrical.

1 17. The channel of claim 15 wherein the first wall further comprises a first radius and
2 the second wall further comprises a second radius.

- 1 18. The channel of claim 17 wherein the first radius is less than the second radius.
- 1 19. The channel of claim 14 further comprising at least one splitter located at least
2 partially within the third space.
- 1 20. The channel of claim 14 wherein the first heat reservoir comprises a copper
2 material configured as a heating element.
- 1 21. The channel of claim 14 wherein the second heat reservoir comprises a copper
2 material configured as a heat sink.
- 1 22. A channel for separating particles comprising:
 - 2 (a) a first heat reservoir means for bearing a first temperature;
 - 3 (b) a second heat reservoir means for bearing a second temperature; and
 - 4 (c) a heat conductive means for bearing a range of temperatures between the
5 first and second temperatures that allows for the development of a thermal diffusion
6 force.
- 1 23. The channel of claim 22 wherein the heat conductive means comprises a fluid
2 means.
- 1 24. The channel of claim 22 wherein the first temperature is less than the second
2 temperature.
- 1 25. A channel for separating particles comprising:
 - 2 (a) a first cylindrical wall having a first radius and a first electrical potential;
 - 3 (b) a second cylindrical wall having a second radius and a second electrical
4 potential, the second radius being greater than the first radius and the second electrical
5 potential being greater than the first electrical potential; and
 - 6 (c) a space between the first and second walls and configured to allow
7 development of an electrophoretic force.

1 26. The channel of claim 25 further comprising a splitter located at least partially
2 within the space.

1 27. A method of separating particles comprising the steps of:
2 (a) generating a first electrical potential;
3 (b) generating a second electrical potential;
4 (c) continuously introducing a plurality of particles into a space between the
5 first electrical potential and the second electrical potential;
6 (d) subjecting the particles to a radial electrophoretic force causing at least
7 some of the particles to undergo radial displacement; and
8 (e) continuously recovering at least a portion of the particles.

1 28. The method of claim 27 wherein the first electrical potential is a varying electrical
2 potential.

1 29. The method of claim 27 wherein the second electrical potential is a varying
2 electrical potential.

1 30. A method for separating particles comprising the steps of:
2 (a) generating an organic liquid phase flow;
3 (b) generating an aqueous liquid phase flow;
4 (c) continuously introducing a mixture of polar and non-polar particles into
5 the aqueous liquid phase flow thereby causing at least a portion of the non-polar particles
6 to move from the aqueous liquid phase flow to the organic liquid phase flow; and
7 (d) continuously recovering at least a fraction of the polar and non-polar
8 particles from the organic liquid phase flow and the aqueous liquid phase flow.

1 31. A channel for separating particles comprising:
2 (a) a first cylindrical wall having a first radius;

- 3 (b) a first plurality of electrodes interdigitated along at least a portion of the
4 first cylindrical wall;
5 (c) a second cylindrical wall having a second radius, the second radius being
6 greater than the first radius;
7 (d) a second plurality of electrodes interdigitated along at least a portion of the
8 second cylindrical wall; and
9 (e) a space between the first and second cylindrical walls configured to allow
10 development of a dielectrophoretic force.

1 32. The channel of claim 31 wherein the first plurality of electrodes comprises a
2 plurality of electrodes each configured circumferentially on the inner cylindrical wall.

1 33. The channel of claim 31 wherein the second plurality of electrodes comprises a
2 plurality of electrodes each configured circumferentially on the outer cylindrical wall.

1/3

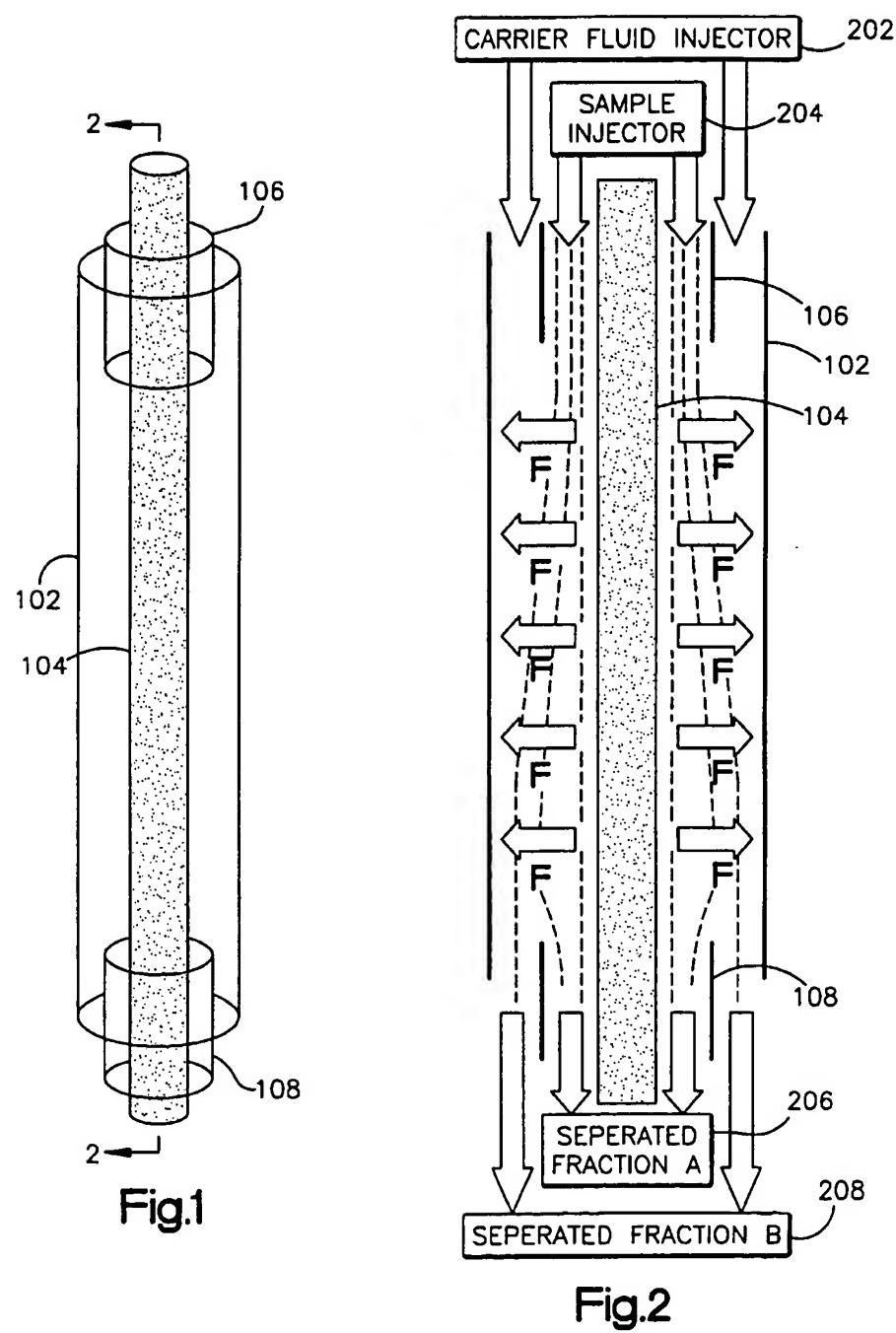
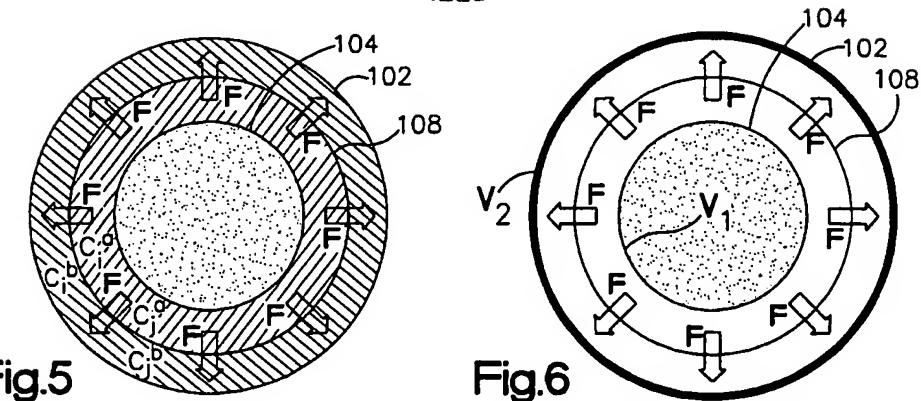
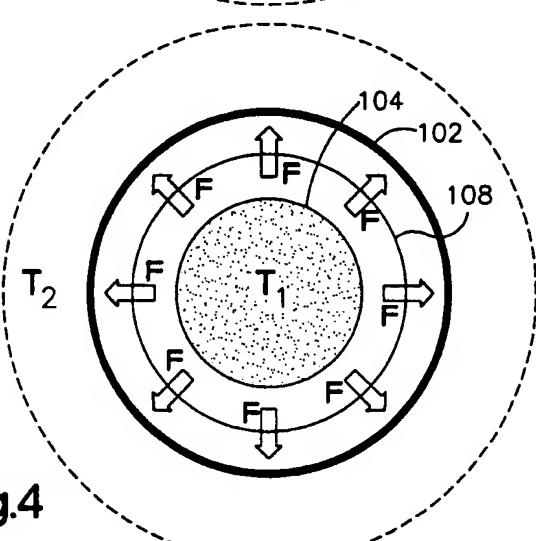
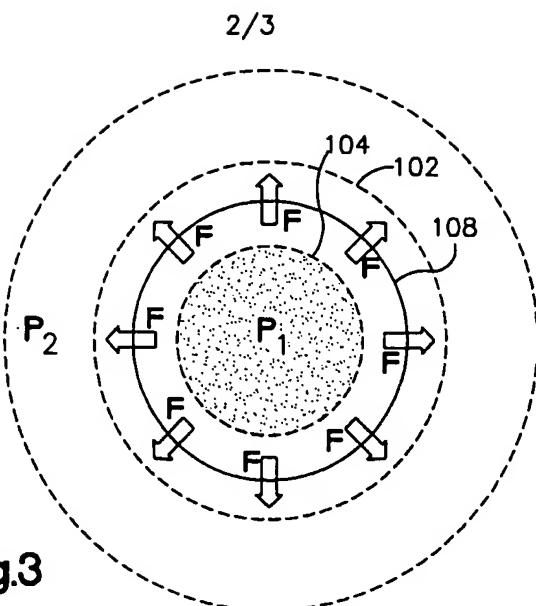


Fig.1

Fig.2



3/3

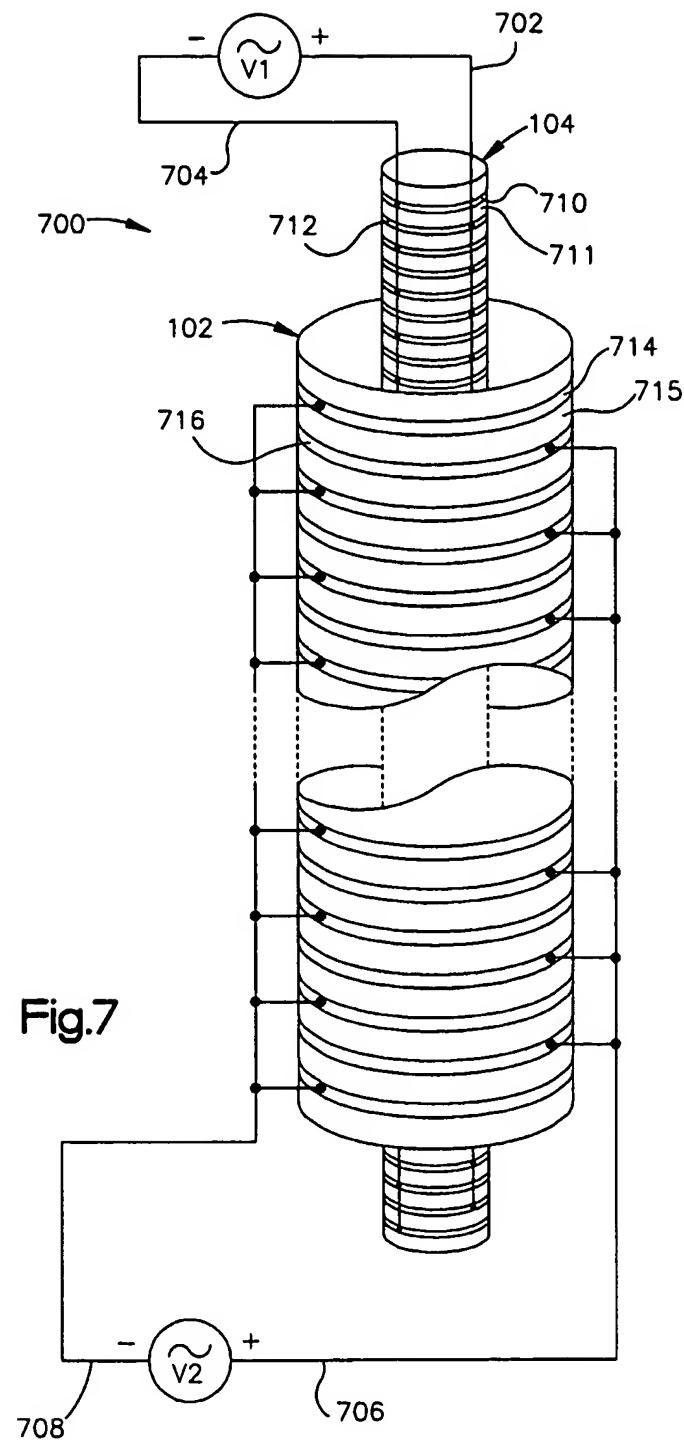


Fig.7